

Oxalic Acid

SDS

1. Product Identification

Synonyms: Ethanedioic acid; Oxalic acid

| | | |
|------------------|-----------|--------------|
| | Anhydrous | Dihydrate |
| CAS No | 144-62-7 | 6153-56-6 |
| Molecular Weight | 90.03 | 126.07 |
| Chemical Formula | HOCCOOH | HOCCOOH.2H2O |



Supplier: Chemical Store Inc. (ChemicalStore.com)
1059 Main Avenue, Clifton, NJ 07011
Phone: (973) 405-6248

2. Hazards Identification

Emergency Overview:

POISON! DANGER! MAY BE FATAL IF SWALLOWED. CORROSIVE. CAUSES SEVERE IRRITATION AND BURNS TO SKIN, EYES, AND RESPIRATORY TRACT. HARMFUL IF INHALED OR ABSORBED THROUGH SKIN. MAY CAUSE KIDNEY DAMAGE.

OSHA Hazards

Target Organ Effect, Harmful by ingestion., Harmful by skin absorption., Irritant, Teratogen

Target Organs

Kidney, Nerves, Blood, Eyes

GHS Classification

Acute toxicity, Oral (Category 4)
Acute toxicity, Dermal (Category 4)
Serious eye damage (Category 1)

GHS Label elements, including precautionary statements

Pictogram

Signal word Danger

Hazard statement(s)

H302 + H312 Harmful if swallowed or in contact with skin
H318 Causes serious eye damage.

Precautionary statement(s)

P280 Wear protective gloves/ eye protection/ face protection.
P305 + P351 + P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

HMIS Classification

Health hazard: 2
Chronic Health Hazard: *
Flammability: 0
Physical hazards: 0

NFPA Rating

Health hazard: 2
Fire: 0
Reactivity Hazard: 0

Potential Health Effects

Inhalation: May be harmful if inhaled. Causes respiratory tract irritation.

Skin: Harmful if absorbed through skin. Causes skin irritation.



Eyes: Causes eye irritation.
Ingestion: Harmful if swallowed.

3. Composition/Information on Ingredients

Formula: $C_2H_2O_4$

Molecular Weight: 90.03 g/cm³

| Ingredient | CAS No | EC-No. | Percent | Hazardous |
|-------------|----------|-----------|-----------|-----------|
| Oxalic Acid | 144-62-7 | 205-634-3 | 99 - 100% | Yes |

4. First Aid Measures

Inhalation: Remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician immediately.

Ingestion: DO NOT INDUCE VOMITING! Give large quantities of limewater or milk to drink. Never give anything by mouth to an unconscious person. Call a physician immediately.

Skin Contact: In case of contact, wipe off excess from skin then immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Wash clothing before reuse. Call a physician immediately.

Eye Contact: Immediately flush eyes with gentle but large stream of water for at least 15 minutes, lifting lower and upper eyelids occasionally. Call a physician immediately.

5. Fire Fighting Measures

Fire: Oxalic Acid is a combustible solid below 101C (215F)

Explosion: Reacts explosively with strong oxidizing materials and some silver compounds.

Fire Extinguishing Media: Water spray, dry chemical, alcohol foam, or carbon dioxide. Foam or water on molten oxalic acid may cause frothing. Water spray may be used to keep fire exposed containers cool.

Special Information: In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with full facepiece operated in the pressure demand or other positive pressure mode.

6. Accidental Release Measures

Remove all sources of ignition. Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8.

Spills: Clean up spills in a manner that does not disperse dust into the air. Use non-sparking tools and equipment. Pick up spill for recovery or disposal and place in a closed container. Remove unnecessary people. If material comes in contact with water, neutralize liquid with alkaline material (soda ash, lime), then absorb with an inert material (e.g. vermiculite, dry sand, earth) and place in a chemical waste container. Do not use combustible materials, such as saw dust. Do not flush to sewer.

7. Handling and Storage

Keep in a tightly closed container. Protect from physical damage. Store in a cool, dry, ventilated area away from sources of heat, moisture and incompatibilities. Containers of this material may be hazardous when empty since they retain product residues (dust, solids); observe all warnings and precautions listed for the product.

8. Exposure Controls/Personal Protection**Airborne Exposure Limits:**

-ACGIH Threshold Limit Value (TLV) : 1 mg/m³ (TWA), 2 mg/m³ (STEL)

-OSHA Permissible Exposure Limit (PEL): 1 mg/m³

Ventilation System:

A system of local and/or general exhaust is recommended to keep employee exposures below the Airborne Exposure Limits. Local exhaust ventilation is generally preferred because it can control the emissions of the contaminant at its source, preventing dispersion of it into the general work area. Please refer to the ACGIH document, Industrial Ventilation, A Manual of Recommended Practices, most recent edition, for details.

Personal Respirators (NIOSH Approved): If the exposure limit is exceeded, a half-face respirator with an organic vapor cartridge and dust/mist filter may be worn for up to ten times the exposure limit or the maximum use concentration specified by the appropriate regulatory agency or respirator supplier, whichever is lowest. A full-face piece respirator with an organic vapor cartridge and dust/mist filter may be worn up to 50 times the exposure limit, or the maximum use concentration specified by the appropriate regulatory agency or respirator supplier, whichever is lowest. For emergencies or instances where the exposure levels are not known, use a full-face piece positive-pressure, air-supplied respirator. **WARNING:** Air-purifying respirators do not protect workers in oxygen-deficient atmospheres.

Skin Protection: Wear protective clothing.

Eye Protection: Use chemical safety goggles and/or full face shield where dusting or splashing of solutions is possible. Maintain eye wash fountain and quick-drench facilities in work area.

9. Physical and Chemical Properties

Appearance: Transparent, colorless crystals.

Solubility: ca. 1g/7mL of water.

% Volatiles by volume @ 21C (70F): 0

Boiling Point: 149 - 160C (300 - 320F) Sublimes.

Vapor Pressure (mm Hg): < 0.001 @ 20C (68F)

Evaporation Rate (BuAc=1): No information found.

Odor: Odorless.

pH: No information found.

Specific Gravity: 1.65 @ 18.5C/4C

Melting Point: 101.5C (216F)

Vapor Density (Air=1): 4.4

10. Stability and Reactivity

Stability: Stable under ordinary conditions of use and storage. Heat will contribute to instability.

Hazardous Decomposition Products: Carbon dioxide and carbon monoxide may form when heated to decomposition. May also form formic acid.

Hazardous Polymerization: Will not occur.

Incompatibilities: Alkalis, chlorites, hypochlorites, oxidizing agents, furfuryl alcohol and silver compounds.

Conditions to Avoid: Heat, ignition sources and incompatibilities.

11. Toxicological Information

Oral rat LD50: 375 mg/kg; irritation skin rabbit: 500 mg/24H mild; eye rabbit 250 ug/24H severe; investigated as a reproductive effector.

Cancer Lists: No

12. Ecological Information

Environmental Fate: No information found.

Environmental Toxicity: No information found.

13. Disposal Considerations

Whatever cannot be saved for recovery or recycling should be managed in an appropriate and approved waste facility. Although not a listed RCRA hazardous waste, this material may exhibit one or more characteristics of a hazardous waste and require appropriate analysis to determine specific disposal requirements. Processing, use or contamination of this product may change the waste management options. State and local disposal regulations may differ from federal disposal regulations. Dispose of container and unused contents in accordance with federal, state and local requirements.

14. Transport Information

Domestic (Land, D.O.T.)

Proper Shipping Name:

CORROSIVE, SOLID, ACIDIC, ORGANIC N.O.S. (OXALIC ACID, DIHYDRATE)

Hazard Class: 8, UN/NA: UN3261, Packing Group: III

International (Water, I.M.O.)

Proper Shipping Name:

CORROSIVE, SOLID, ACIDIC, ORGANIC N.O.S. (OXALIC ACID, DIHYDRATE)

Hazard Class: 8, UN/NA: UN3261, Packing Group: III

Information reported for product/size: 12KG

15. Regulatory Information

Federal and State Regulations:

Illinois toxic substances disclosure to employee act: Oxalic acid anhydrous Rhode Island

RTK hazardous substances: Oxalic acid anhydrous Pennsylvania RTK: Oxalic acid

anhydrous Minnesota: Oxalic acid anhydrous Massachusetts RTK: Oxalic acid anhydrous

New Jersey: Oxalic acid anhydrous California Director's list of hazardous substances:

Oxalic acid anhydrous TSCA 8(b) inventory: Oxalic acid anhydrous

Other Regulations:

OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).

EINECS: This product is on the European Inventory of Existing Commercial Chemical Substances.

Other Classifications:

WHMIS (Canada): CLASS E: Corrosive solid.

DSCL (EEC): R21/22- Harmful in contact with skin and if swallowed. S24/25- Avoid contact with skin and eyes.

16. Other Information

Label Hazard Warning:

POISON! DANGER! MAY BE FATAL IF SWALLOWED. CORROSIVE. CAUSES SEVERE IRRITATION AND BURNS TO SKIN, EYES, AND RESPIRATORY TRACT. HARMFUL IF INHALED OR ABSORBED THROUGH SKIN. MAY CAUSE KIDNEY DAMAGE.

Label Precautions: Do not breathe dust. Do not get in eyes, on skin, or on clothing. Keep container closed. Use only with adequate ventilation. Wash thoroughly after handling.

Label First Aid: If swallowed, DO NOT INDUCE VOMITING! Give large quantities of lime water or milk to drink. Never give anything by mouth to an unconscious person. If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. In case of contact, wipe off excess material from skin then immediately flush eyes or skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Wash clothing before reuse. In all cases call a physician immediately.

Product Use: Laboratory Reagent; Automobile radiator cleaner; general metal and equipment cleaning; leather tanning; catalyst; stripping agent for permanent press resins; bleaching of textiles.

Revision Information: GHS Pictograms added.

Disclaimer:

The information above is believed to be accurate and represents the best information currently available to us. However, we make no warranty of merchantability or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes. In no event shall Chemical Store Inc be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential or exemplary damages, howsoever arising, even if Chemical Store Inc has been advised of the possibility of such damages.